

2.11 30320 ORGANIC CHEMISTRY

UNIT-1 FUNDAMENTALS OF ORGANIC CHEMISTRY

- 1.1 Introduction of organic chemistry
- 1.2 Modern concept of organic chemistry
- 1.3 Scope of organic chemistry as separate branch of chemistry
- 1.4 Distinguish between organic compounds and inorganic compounds
- 1.5 Classification:
 - a) Based on the nature of carbon chain
 - b) Nature of chemical reactions
 - c) Nature of functional groups
- 1.6 Hydrocarbons
- 1.7 Saturated hydrocarbons – examples
- 1.8 Unsaturated hydrocarbons – examples
- 1.9 Homologous series and homologues
- 1.10 Functional groups and its importance
- 1.11 Definition of Isomerism – Isomers
- 1.12 Classification of isomerism
- 1.13 Structural isomerism – Chain – Position- functional – Metamerism – examples for each.
- 1.14 Stereoisomers – Geometrical – Cis-trans isomers – examples – optical isomers (definition)
- 1.15 Organic reactions definition and examples
- 1.16 Types of organic reactions
 - a) Substitution of Methane with Chlorine
 - b) Addition reaction – Hydrogenation of Ethene and Ethyne - Chlorination of Ethene and Ethyne
 - c) Markownikoff's rule
 - d) Elimination reaction – condensation reaction of formaldehyde – Ketones
- 1.17 Rearrangement reaction
- 1.18 Polymerisation reaction – Polyethylene PVC, TEFLON, NYLONS, POLYESTERS
- 1.19 Reduction reactions
- 1.20 Oxidation reactions
- 1.21 Biochemical reactions
- 1.22 Fermentation reactions

UNIT-2 REACTION MECHANISM

- 2.1 Homolytic fission
- 2.2 Etrolytic fission
- 2.3 Carboniumions
- 2.4 Carbanian
- 2.5 Electrophilic reactions
- 2.6 Nucleophilic reactions
- 2.7 Electron displacement effects
- 2.8 Inductive effects

- 2.9 Electromeric effect
- 2.10 Free radicals
- 2.11 Nomenclature of organic compounds
- 2.12 Common or trivial system
- 2.13 IUPAC system
- 2.14 IUPAC rules for naming organic compounds

UNIT-3 GENERAL STUDY OF STRUCTURAL FORMULA, PREPARATION/MANUFACTURE, PROPERTIES AND USES OF THE FOLLOWING HYDROCARBONS:

- 3.1 Methane
- 3.2 Ethene & Ethyne
- 3.3 Benzene
- 3.4 Phenols – Phenol
- 3.5 Ethers – Diethyl ether
- 3.6 Aldehydes – Formaldehyde
- 3.7 Ketanes – Acetone
- 3.8 Carboxylic Acids – Formic Acid
- 3.9 Esters – Ethylacetate
- 3.10 Amines – Methylamine

UNIT-4 ORGANIC COMPOUNDS OF BIOLOGICAL IMPORTANCE. (BRIEF STUDY OF MOLECULAR FORMULA, STRUCTURAL FORMULA, SOURCES, CLASSIFICATION, PROPERTIES AND USES OF):

- 4.1 Carbohydrates
- 4.2 Glucose & Sucrose
- 4.3 Starch & Cellulose
- 4.4 Proteins & Hormones
- 4.5 Insuline
- 4.6 Oxytocin
- 4.7 Adrenaline
- 4.8 Vitamins – A, B,(complex) C,D,E and K
- 4.9 Vitamin deficiency diseases
- 4.10 Food flavors – Natural and Synthetic
- 4.11 Perfumes – Natural & Synthetics
- 4.12 Drugs – Characteristics – Analgesic – Antipyretics – Antibiotics
- 4.13 Dyes – Natural and synthetic dyes

Reference Books:

- 1. T.B of Organic Chemistry - Arun Bahl, B.S. Bahl
- 2. Organic Chemistry - Solomons & Fryhle
- 3. Organic Chemistry Vol-I & II - I.L. Finar
- 4. Pharmaceutical Chemistry (Organic) - G.R. Chathval